Flame Retardants, PFAS and Firefighters



Plus ça change, plus la même chose...



The James and Jeanette Neckers Lectureship in Chemistry

Winants Auditorium, Graves Hall and the DeWitt Theatre March 10-11, 2011 Hope College

Dr. Arlene Blum

Executive Director of the Green Science Policy Institute and Visiting Scholar in the Department of Chemistry University of California, Berkeley

> Public Lecture Seminar 7:00 pm DeWitt Theatre

Climbing Your Own Everests: Peaks, Public Health, and Policy

Cimbing the world's highest mountains is an excellent model for activating other extremely demanding objectives. Dr. Blum will share photos and stones from the recently and policy work protecting our health and environment from toxins in consumer products, as well as from the expectitions among the world's highest and most dangerous mountains. Dr. Blum's current challenge is bringing scientists, industry, government, and non-profits together to create a healtheir, safer environment. Her work bringing science is not regulatory decisions has stopped the nuncessary use of hundreds of millions of founds of toxic chemicals.

Tris(1,3-dichloro-2-propyl)
phosphate
Flame retardant in
common use today



Tris(2,3-dibromopropyl)
phosphate
Flame retardant banned in 1977
for use in children's clothing

Neckers Lecture Refreshments 3:30, Seminar 4:00 Winants Auditorium, Graves Hall

Organohalogens in Consumer Products: Do the Fire Safety Benefits Justify the Health and

Organolalogens are commonly used as flame retardants in consumer products, particularly firmitine and baby products, although it is not clear that these compounds really lead to increesed fire safety. In addition, some organohalogens are known to cause neurological, reproductive, and thryoid health problems, as well as cancer. As the products move to landfills and other waste streams, they also negatively impact the environment. In this talk, Dr. Blum will discuss how academic scientifics can be engaged with policy making to stop the unnecessary use of potentially toxic flame retradunts in consumer products.

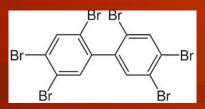


Dr. Adme Blum received her bachelor's degree from Reed College and her Ph.D. in biophysical chemistry from the University of California, Berkeley, where she worked under the mentorship of Dr. Ignation Timcoo. As a post-doe with Dr. Bruce Ames at Berkeley, Dr. Blum demonstrated the cancer-cassing properties of two flume retardants whelly used in children's sleepware, which led directly to their regulation (for example, the brommated molecule, above). Dr. Blum took a break flom science to build a reputation as a mountain climber and author. She led the first American—and all-women 's—secret of Anuapuran I (8091m), considered one of the world's nost dangerous and difficult mountains. She also led the first women's team up Mt. McKinley (6,194m) and was the first American woman ot attempt Mt. Everset (8,484m). After three decades of climbing and writing books, Dr. Blum returned to science and policy work in 2006 after learning that the same chemicals she worked to remove from children's sleepware in the 1970's are currently used in firmitate and baby products. Through the Green Science Policy Institute she seeks to bring scientific research to the Torefort to to inform policy decisions with regard to human Bealth

and environmental impact. For her work, Dr. Blam has received a number of awards including being named as one of the 100 "Fount Tailing by the Leaf to Sirve Our Planet" if the Purpose Pizze, given to those over 60 who are solving society's greatest problems, and a Gold Medal from the Society of Women Geographers, an homor previously given to only eight of their women, including Amelia Enharth, Margaret Meda, and Mary Leakey.

In addition to the Chemistry Department's Neckers Fund, Dr. Blum's visit is co-sponsored by the Campus Sustainability Task Force, the Hope College HHMI grant, Geology and Environmental Sciences, and Women's Studies.

- 9 years ago...
- Flame retardants...



Do something...



Michigan Michigan



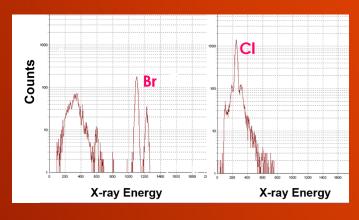
PIXE For Flame Retardant Detection





flame retardants: by detecting Br, Cl





Science & Policy





Playing with fire

A deceptive campaign by industry brought toxic flame retardants into our homes and into our bodies. And the chemicals don't even work as promised.



By PATRICEA CALAMAN AND SAM ROE

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doctors, environmentalists and even firefighters sound abstract and petty.

But there was a problem with his testimony: It wasn't true.

Records show there was no dangement pillow or candle fire. The holy be described didn't crisit.

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Edmund G. Brown Jr.

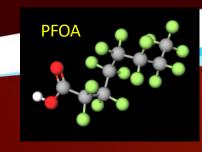


FOR IMMEDIATE RELEASE:

Monday, June 18, 2012

Governor Brown Directs State
Agencies to Revise Flammability
Standards

What makes PFAS so good?



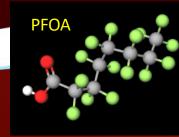
These compounds are used everywhere:



Average Bond Energies (kJ/mol)

	Bond	Energy	
Single Bonds			
	H-H	432	
	H-F	565	
	H-Cl	427	
	H—Br	363	
	H-I	295	
	C-H	413	
	C-C	347	
	C—Si	301	
	C-N	305	
	c-o	358	
	C—P	264	
	C-S	259	
	C-F	453	
	C—CI	339	
	C—Br	276	
	C—I	216	

So what is the problem?



The carbon-fluorine bond is very hard to break-these compounds have environmental lifetimes ≈100's of years or more!

Many have no known biotic or abiotic degradation pathways....

ENVIRONMENTAL PERSISTENCEThe "forever chemicals" ...

Average Bond Energies (kJ/mol)

Bond Energy

Single Bonds

H—H 432

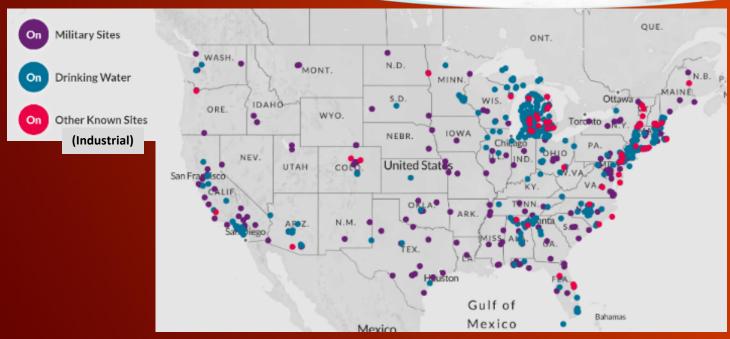
H—F 565

H—Cl 427

H—Br 363

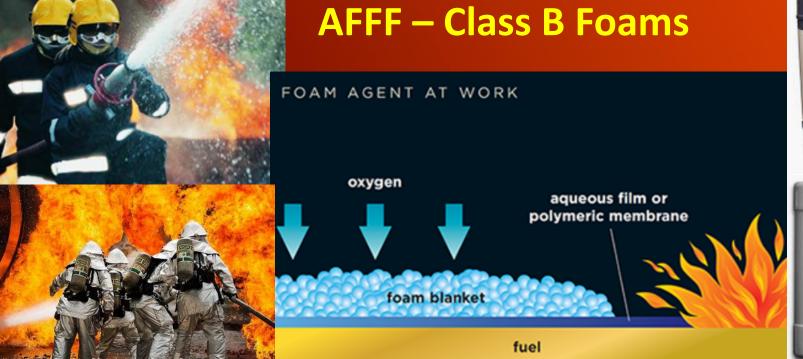
H-I295 413 C-HC-C347 C-Si 301 C-N305 C-O358 C-P264 C-S259 453 339 C-Br 276 C-I216

These are only the US sites to date...



https://www.ewg.org/interactive-maps/2019_pfas_contamination/map/

PFAS and Firefighters







DoD extensive use...







A U.S. Air Force airman blows firefighting foam that was unintentionally released in an aircraft hangar at Travis Air Force Base, Calif., in 2013. Toxic chemicals in the foam have contaminated private wells in Box Elder and municipal wells in Sioux Falls.



AFFF – Class B Foams



PFAS Exposure sources:

Active fire event

Decon post event

Aerosolized foam & combustion products

Drinking water contamination***

Are there other PFAS exposure sources?



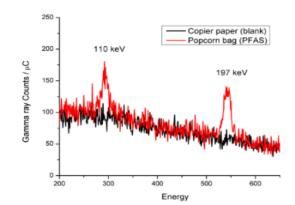


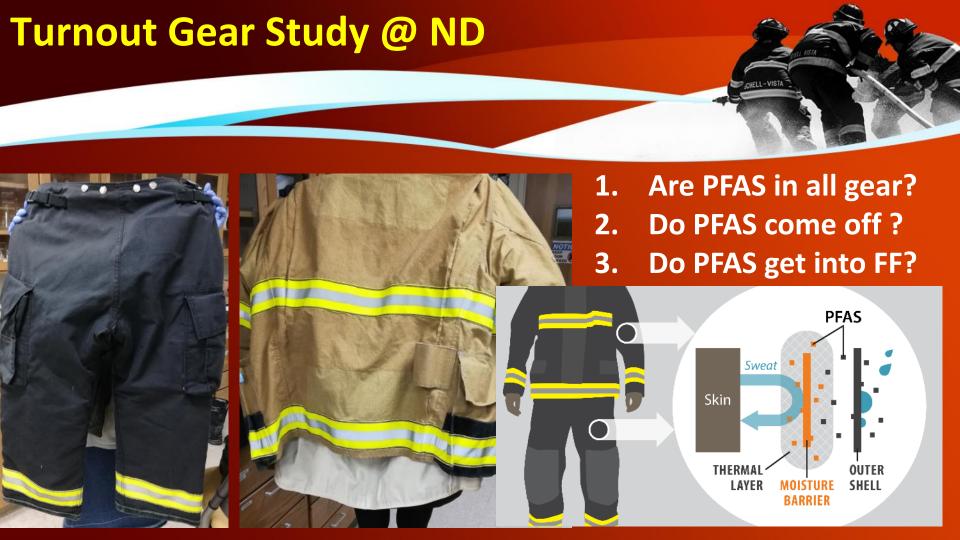


PIGE Analysis of Fluorine



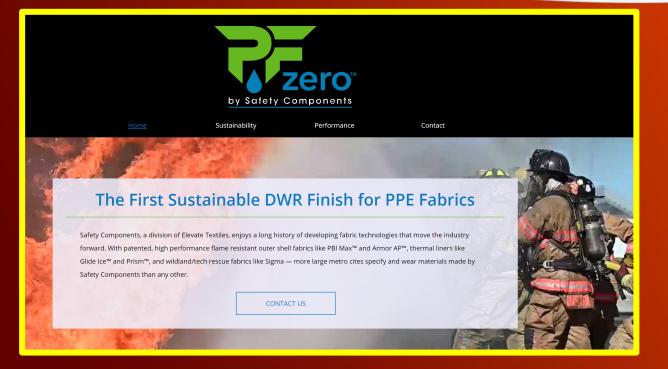






Science & Policy





January, 2020

Next: Commercialization of PIGE



Partner with medical cyclotron manufacturer:

Use GENTrace cyclotron (100,000 samples/yr)
 Run 3 L of water through GAC felt filter



(minutes in field)

Dry filters (hours)

Send filters for PIGE analysis

(1-2 minutes per sample)

Identify all sum of all PFAS present

Screening method: No F, no PFAS....



Acknowledgements



John Wilkinson, Sean McGuinness, Meghanne Tighe, Ashabari Majumdar, Nick Caterisano, Matt Roddy, Alec Gonzales, Seryoeng Lee, Chase Miller





